

Organic bio-electronic sensors for ultra-sensitive chiral differential detection



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- 2 printable bio-electronic field-effect transistors
- 3 detections down to fM, chiral differential detection with an ESF > 6
- 4 Odorant Binding and anti-C Reactive Protein as cases of study



printable electronics





minority report - Steven Spielberg (2002)







SAMSUNG Flexible AM OLED





point-of-care biosensors



sensing arrays: this is the way to go



Affymax DNA chip







http://bmel.korea.ac.kr/image/intro_fig_5_4.gif





optical and electrochemical method

PROS

- Easy miniaturization
- CMOS compatible (no reference electrode) - Label-free
- Low cost (printing fab on plastic, fabric, paper)

Y

CONS

- Totally novel approach
 - New production paradigms
- Critical is the control of the interfaces

the search for quantitative stick testing



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Schematic drawing of the RenaStick dipstick showing advantages and disadvantages of point-of-care testing in the setting of acute kidney injury. www.nature.com 2014

printed circuits on paper



electronic OFET bio-sensors @UNIBA

DIELECTRIC

osc

OSC

DIELECTRIC G

Chem Soc Rev

TUTORIAL REVIEW

OFET

bio-sensors

SVYYYYD

DIELECTRIC

Chemical Society Reviews

Organic field-effect transistor sensors: a tutorial

Luisa Torsi,* Maria Magliulo, Kyriaki Manoli and Gerardo Palazzo

Organic thin-film transistors embedding biological recognition elements are successfully employed as ultrasensitive, low-cost, label-free biosensors in several analytical fields.

L. Torsi et al. Chem. Soc. Rev., 2013, 42 (22), 8612 - 8628

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RSCPublishing

c3cs60127g

impact of a binding event on electronic properties



V_T changes

 \Box impacts on the OSC transport properties $\rightarrow \mu_{FET}$ changes

 $\Box \text{ Impacts on the gating system capacitance} \rightarrow C_i \text{ changes}$

Ion Selective FET (ISFET) like bio-sensor



GAS & Ion-Selective FET (ISFET)



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L. Torsi et al Angew. Chem. Int. Ed. 2015, 54, 12562-12576

FBI-OFET sensors

April 24, 2012 | vol. 109 | no. 17 | pp. 6429-6434

Proceedings of the National Academy of Sciences of the United States of America

www.pnas.org

Interfacial electronic effects in functional biolayers integrated into organic field-effect transistors

electronic transduction of proton translocation

ADVANCED

www.advmat.de



www.MaterialsViews.com

Tailoring Functional Interlayers in Organic Field-Effect Transistor Biosensors

Maria Magliulo, Kyriaki Manoli, Eleonora Macchia, Gerardo Palazzo, and Luisa Torsi*



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L. Torsi et al Angew. Chem. Int. Ed. 2015, 54, 12562-12576

how Electrolyte Gated OFETs work



water self-ionization

H. Klauk, Organic electronics II: More materials and applications; Wiley-VCH S. H. Kim, K. Hong, W. Xie, K. H. Lee, S. Zhang, T. P. Lodge, C. D. Frisbie, Adv. Mater. 2013, 25, 1822-1846;

DEGLI STUDI DI BARI

Polyelectrolytes in EGOFET ACS APPLIED MATERIALS & INTERFACES **Research Article**

www.acsami.org

Plain Poly(acrylic acid) Gated Organic Field-Effect Transistors on a **Flexible Substrate**

Liviu M. Dumitru, Kyriaki Manoli, Maria Magliulo, Luigia Sabbatini, Gerardo Palazzo, and Luisa Torsi* Department of Chemistry, "Aldo Moro" University, Via Orabona 4, Bari 70126, Italy



EGOFET - architectures



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New polyelectrolyte as gating material in OFETs





Calcium alginate fruit (blueberry) "caviar"



Alginate capsules



Electrical performance



Green Electronics



M. Irimia-Vladu et la. Adv. Funct. Mater., 2010, 20 C. J. Bettinger and Z. Bao , Adv. Mater., 2010 , 22 , 651..

Printing pigments





ADVANCED MATERIALS

www.MaterialsViews.com

www.advmat.de

Hydrogen-Bonded Semiconducting Pigments for Air-Stable Field-Effect Transistors

Eric Daniel Głowacki,* Mihai Irimia-Vladu, Martin Kaltenbrunner, Jacek Gąsiorowski, Matthew S. White, Uwe Monkowius, Giuseppe Romanazzi, Gian Paolo Suranna, Piero Mastrorilli, Tsuyoshi Sekitani, Siegfried Bauer, Takao Someya, Luisa Torsi, and Niyazi Serdar Sariciftci



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From kitchen to lab: Curiosity driven research



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how EGOFETs work



water self-ionization

H. Klauk, Organic electronics II: More materials and applications; Wiley-VCH S. H. Kim, K. Hong, W. Xie, K. H. Lee, S. Zhang, T. P. Lodge, C. D. Frisbie, Adv. Mater. 2013, 25, 1822-1846;

the capacity modulated device



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electrolyte gated OFET (EGOFET) sensor



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water gated OFET sensors - review





charged species → stronger long-range coulomb interactions (10-100 kJ/mol) → electrochemical potential → ISFET

neutral species (or species carrying a dipole moment) → weaker shortrange interactions such as the dipole-dipole or the dispersive ones (2 kJ/mol) → chemical potential → capacity modulated FET

L. Torsi et al Angew. Chem. Int. Ed. 2015, 54, 12562-12576

Odorant Binding Proteins EGOFET sensor



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Odorant Binding Proteins



Odorant Binding Proteins



Shuttle odorant molecules Odorant clearance mechanism

FLEXSmell

Carton Pack **BO**test

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vertebrate odorant system



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why odorant binding proteins ?

- OBPs are present in high concentrations of mM range in mammalian nose and insects antennae
 - Soluble proteins, can be expressed in bacterial systems at low-cost
- Highly stable in ambient/hot conditions
 - Binds reversibly to odorants



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COLI POLYTICHNIQUI





MANCHESTER









pig Odorant Binding Proteins (pOBP)

Pig odorant binding proteins (pOBP)



The protein is characterized by a hydrophobic β -barrel cavity,

Differently from other OBPs such as the bovine one, pOBP b-barrel cavity is devoid of naturally occurring bound ligand

It bears a negative charge

No study on chiral interactions; carvone enantiomers





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OUL FOUTICHNIOU



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chiral ligand molecules



The binding process



competitive fluorescent biding assay



OBP in a water gated **OFET** sensor





Cinzia Di Franco - CNR - Bari Maria Vittoria Santacroce and Gaetano Scamarcio - University of Bari

Mulla, M.Y. et al. Nature Communications, 2015, 6, 6010

Electrochemical pOBP SAM characterization

Hindrance (B) 0.6 MANCHESTER Au-3MPA (SAM) 0.4 -10 0.2 -20 Au-Bare -30 0.0 -0.2-0.1 0.0 0.1 0.2 0.3 0.4 0.5 0.6 0.7 Electrode E (V) vs Ag/AgCl $B = 1 - (I_0^{Fun} / I_0^{Au})$ Where, I_0^{Fun} and I_0^{Au} are the oxidative peak currents obtained from the CV curves Holst Centre for functionalized electrode (3MPA alone and 3MPA-pOBP) and the bare Au electrode respectively Carton Pac

K₄Fe(CN)₆ electro-active

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FLEXSmell

Botest

XPS pOBP SAM characterization



I-V Characteristics



 $V_T = -0.01 \pm 0.06$ V, *on/off* current ratio of 150 ± 110 and $\mu_{\text{FET}} = 1.1 \pm 0.2 \ 10^{-1} \text{ cm}^2/\text{Vs.}$

Sensing of tine neutral species with WGOFETs **FLEXSmell** UNIVERSITÀ DEGLI STUDI DI BARI ALDO MORO -1,6**pOBP-SAM** COLI POLYTICHNIQUI MEALE DE LAUSAN ,100 pM **pOBP WGOFET exposed to** nc -1,2 **500 рМ** (S)-(+)-carvone MANCHESTER Vn-0,8 S l nM .5 nM -0,4 VT 0,0 Holst Centre -0,1 -0,2 -0,3 -0,5 0,0 -0,4 Carton Pack Botest

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member of the ASVS G

dose-curves with WGOFETs



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the two enantiomers on the very same gate



Decoupling capacitance and threshold voltage

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The response is dominated by capacitance



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(S)-(+)carvone / pOBP complex



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electronic OFET Bilayer configuration



A sensitivity-enhanced field-effect chiral sensor

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Figure 1 Bilayer OTFT chiral sensor structure. The transistor has a bottom-gate device structure that consists of a highly n-doped silicon wafer (resistivity 0.02–1 Ω cm⁻¹) nature materials IVOL 7 I MAY 2008 I www.nature.com/naturematerials

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(R)-(-)carvone / pOBP complex



enanthio-selectivity factor > 6



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markedly different levels of cooperativity and an exceptionally high ESF

The pOBP capacitance model



2.4 % of the total surface of the protein exposed to the solvent (~ 7 nm²) it well accommodate a few water molecules



G. Lattanzi

R(-)-Carvone \longrightarrow $C_{OBP} \simeq C_w$ **S(+)-Carvone** \longrightarrow $C_{OBP} = C_p$

high dielectric percolative path, "water channel"

dissociation constant for 2D receptors



 $\Delta G^0 = \operatorname{RT} \ln[K_{\mathbf{D}}]$ **Surface segregated OBP** $\Delta G^{\circ}_{FET}^{(+)} = - (49.2 \pm 0.1) \text{ kJ/mol}$ $\Delta G^{\circ}_{FET}^{(-)} = -(41 \pm 2) \text{ kJ/mol}$ **OBP** in solution $\Delta G^{\circ}_{sol}^{(+)} = - (36.00 \pm 0.05) \text{ kJ/mol}$ $\Delta G^{\circ}_{sol}(-) = - (33.0 \pm 0.1) \text{ kJ/mol}$



the thermodynamic cycle



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